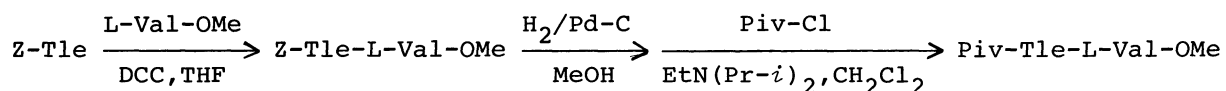


DEPENDENCY OF THE RATIO OF THE DIASTEREOMERS,
L- AND D-*t*-LEUCYL-L-VALINE, ON THE COUPLING METHODS

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In the condensation of *N*-pivaloyl- or *N*-benzyloxycarbonyl-DL-*t*-leucine with methyl L-valinate, the ratio of the resulting diastereomers varied depending upon the coupling method.

In the course of our studies on the asymmetric synthesis of *t*-leucyl peptides,¹⁾ samples of methyl *N*-pivaloyl-*t*-leucyl-L-valinate (Piv-Tle-L-Val-OMe: Tle = DL, D and L) were necessary as standards for the gas chromatographic analysis.²⁾ We attempted to prepare the samples by condensing Z-Tle^{*}) (DL, D and L) with L-Val-OMe by the DCC or the DCC-HOBT method and replacing Z with Piv as shown in the following scheme.



The condensation of Z-DL-Tle with L-Val-OMe, however, gave a product rich in the L-L-diastereomer, particularly by the DCC-HOBT method.

Apparently this phenomenon reflects kinetic difference between the reaction of Z-L-Tle with L-Val-OMe and that of Z-D-Tle with L-Val-OMe. Stimulated by this fact we further examined the diastereomer ratios in this coupling reaction by various methods commonly used.³⁾

Table 1 shows the diastereomer ratios in the products obtained by the coupling of Piv-DL-Tle with L-Val-OMe in the molar ratio of 2:1 at room temperature for *ca.* 24 h. By the DCC method the D-L-isomer was produced preferentially, but the use of such additives as HOBT, HOSu, and HONb⁴⁾ brought about a reverse ratio. Almost the same results were obtained when the active esters of these additives were used. On the other hand by the DPPA method the D-L-isomer was produced in larger excess.⁵⁾

The condensation of Z-DL-Tle with L-Val-OMe (molar ratio = 2:1) was carried out under the same conditions (Table 2). In this case even the condensation by DCC alone produced a mixture rich in the L-L-diastereomer, and the additives increased the L-L-diastereomer content greatly. Especially marked was the result of the DCC-HONb method, the condensation product containing *ca.* 80% of the L-L-diastereomer.

This interesting phenomenon, that the ratio of the L-L- and D-L-diastereomers varies strikingly with the coupling method, may be a special case, and seems to be connected with the bulkiness of both the acid and the amine components. Červinka *et al.*⁶⁾ have also observed a similar phenomenon that in the coupling reaction of Z-DL-amino acids with L-amino acid esters in the molar ratio of 3:2 by the DCC method at

room temperature D-L-diastereomers are preferentially formed, though no reference is made to the other coupling methods.

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Table 1. Condensation of Piv-DL-Tle with L-Val-OMe (2:1)

Method ^{a)}	Yield (%) ^{d)}	Ratio of Diastereomers ^{e)}	
		L-L	D-L
DCC-HONb	70	69	31
Active ester (-ONb)	83	68	32
DCC-HOSu	58	60	40
Active ester (-OBt)	78	60	40
Active ester (-OSu)	45	59	41
DCC-HOBt	88	58	42
MA (ClCOOEt) ^{b)}	-	46	54
MA (ClCOOBu- <i>i</i>) ^{b)}	-	45	55
CDI	-	45	55
Chloride	-	44	56
DCC	89	43	57
Active ester (-ONp)	87	42	58
DPPA ^{c)}	-	31	69

a) Unless otherwise noted THF was used as solvent. b) Solvent: THF-CHCl₃
 c) Solvent: DMF d) The yields, after isolation by a preparative TLC, are overestimated due to contamination. e) Estimated by a Hitachi 063 gas chromatograph: column, 0.5% FFAP on Chromosorb G (φ3mm × 2m); column temp, 150°C; carrier gas, N₂ 11 ml/min.

Table 2 Condensation of Z-DL-Tle with L-Val-OMe (2:1)

Method ^{a)}	Yield (%) ^{b)}	Ratio of Diastereomers ^{b)}	
		L-L	D-L
DCC-HONb	52	81	19
DCC-HOBt	55	73	27
DCC-HOSu	79 ^{c)}	65	35
DCC	79	58	42

a) Solvent: THF b) Estimated after converting the products into Piv-Tle-L-Val-OMe (see the scheme). c) The yield is overestimated due to contamination.

References and Notes

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- * Abbreviations: DCC, dicyclohexylcarbodiimide; Z, benzyloxycarbonyl; HOBt, 1-hydroxybenzotriazole; HOSu, *N*-hydroxysuccinimide; HONb, *N*-hydroxy-5-norbornene-endo-2,3-dicarboximide; MA, mixed anhydride; CDI, carbonyldiimidazole; ONp, *p*-nitrophenyl ester; DPPA, diphenylphosphorylazide.

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